



PATENT APPLICATION

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of

Docket No: Q54488

Masahiro OHMORI, et al.

Appln. No.: 09/579,708 ✓

Group Art Unit: 1754 ✓

Confirmation No.: 7789

Examiner: Steven J. Bos

Filed: May 26, 2000

For: PEROVSKITE TITANIUM-TYPE COMPOSITE OXIDE PARTICLE AND PRODUCTION
PROCESS THEREOF

SUBMISSION OF EXECUTED DECLARATION UNDER 37 C.F.R. §1.132

Commissioner for Patents
Washington, D.C. 20231

Sir:

Submitted herewith is an executed Declaration Under 37 C.F.R. §1.132 signed by
Akihiko Shirakawa and dated September 10, 2003. An undated version of this
executed Declaration was filed with an RCE on October 10, 2003.

Respectfully submitted,

Keiko K. Takagi
Registration No. 47,121

SUGHRUE, MION, ZINN,
MACPEAK & SEAS, PLLC

WASHINGTON OFFICE

23373

CUSTOMER NUMBER

Date: October 20, 2003

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For: PEROVSKITE TITANIUM-TYPE COMPOSITE OXIDE PARTICLE AND PRODUCTION
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DECLARATION UNDER 37 C.F.R. § 1.132

Commissioner for Patents
P.O. Box 1450
Alexandria, VA 22313-1450

Sir:

I, Akihiko Shirakawa, hereby declare and state:

THAT I am a citizen of Japan;

THAT I received a Bachelor's degree from Kyoto University in 1989;

THAT I am employed by Showa Denko K.K., where I hold a position as Assistant
Staff Manager, with responsibility for ceramics research;

THAT in order to show the unexpected superiority of the particles of the
present invention, the following determinations were made by me or under my direct
supervision as described below.

The A/B mole ratio was determined for the Examples and the Comparative
Examples in the present specification.

	Synthesis A/B mole ratio	Product A/B mole ratio
Example 1	0.999	0.999
Example 2	0.998	0.998
Example 3	0.999	0.998
Example 4	1.000	1.000
Comparative Example 1	0.999	0.984
Comparative Example 2	0.998	0.996
Comparative Example 3	1.001	0.975

As shown above, the A/B mole ratio of the product is nearly equal to the A/B mole ratio of the synthesis materials in the Examples of the present invention.

In contrast, the A/B mole ratio of the particles of Bruno is smaller than the A/B mole ratio of the synthesis materials because an alkoxy system is employed in which a washing step is required. See Table 1 in Bruno. In addition, the A/B mole ratio of the particles of Menashi is smaller than the A/B mole ratio of the synthesis materials because a hydrothermal process is employed and the product is washed with water or $\text{Ba}(\text{OH})_2$ solution to reduce the Ba/Ti mole ratio in the product. See col. 7, line 43-57 et seq.

Generally, a washing step generates a great amount of defects when barium compounds are eliminated from the surface of particles. Since both Bruno and Menashi use washing steps, the particles have defects that damage the electrical properties of the particles.

The mole ratio of the particles of the present invention is nearly equal to the mole ratio of the synthesis materials because the process of making the particles of the present invention uses no wash step, which generally generates a great amount of defects when barium compounds are eliminated from the surface of particles. As a result, the particles of the present invention unexpectedly have good electrical properties.

I declare further that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

Date: Sep. 10, 2003

Akihiko Shirakawa
Akihiko Shirakawa